Combined Science - Chemistry - Key Stage 4 Atomic Structure & the Periodic Table

Development of the periodic table

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Periodic Table of Elements

				Key:		_											
1 H hydrogen 1		relo	ative atomi			— Atomic — Atomic	symbol (proton nu	umber)									4 He helium 2
7 Li lithium 3	9 Be beryllium 4					-						11 B boron 5	12 C carbon 6	14 N nitrogen 7	16 O oxygen 8	19 F fluorine 9	20 Ne neon 10
23 Na sodium 11	24 Mg magnesium 12											27 Al aluminium 13	28 Si silicon 14	31 P phosphorus 15	32 S sulfur 16	35.5 Cl chlorine 17	40 Ar argon 18
39	40	45	48	51	52	55	56	59	59	63.5	65	70	73	75	79	80	84
K	Ca	Sc	titanium	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
potassium 19	calcium 20	scandium 21	22	vanadium 23	24	manganese 25	iron 26	cobalt 27	nickel 28	copper 29	zinc 30	gallium 31	germanium 32	arsenic 33	selenium 34	bromine 35	krypton 36
85	88	89	91	93	96	[97]	101	103	106	108	112	115	119	122	128	127	131
Rb	Sr	Y	Zr	Nb	Mo	Тс	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Те		Хе
rubidium 37	strontium 38	yttrium 39	zirconium 40	niobium 41	molybdenum 42	technetium 43	ruthenium 44	rhodium 45	palladium 46	silver 47	cadmium 48	indium 49	tin 50	antimony 51	tellurium 52	iodine 53	xenon 54
133	137	139	178	181	184	186	190	192	195	197	201	204	207	209	[209]	[210]	[222]
Cs	Ba	La*	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	TI	Pb	Bi	Po	At	Rn
caesium	barium	lanthanum	hafnium	tantalum	tungsten	rhenium	osmium	iridium	platinum	gold	mercury	thallium	lead	bismuth	polonium	astatine	radon
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
[223]	[226]	[227]	[267]	[270]	[269]	[270]	[270]	[278]	[281]	[281]	[285]	[286]	[289]	[289]	[293]	[293]	[294]
Fr	Ra	Ac*	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	FI	Mc	Lv	Ts	Og
francium	radium	actinium	rutherfordium	dubnium	seaborgium	bohrium	hassium	meitnerium	darmstadtium	roentgenium	copemicium	nihonium	flerovium	moscovium	livermorium	tennessine	organesson
87	88	89	104	105	106	107	108	109	110	87	112	113	114	115	116	117	118



Independent practice

Write your answers using full sentences:

- 1. What are the rows on the periodic table called?
- 2. What are the columns in the periodic table called?
- 3. What does the group number represent?
- 4. What does the period number represent?
- 5. Completed the sentence: The modern periodic table is ordered using....

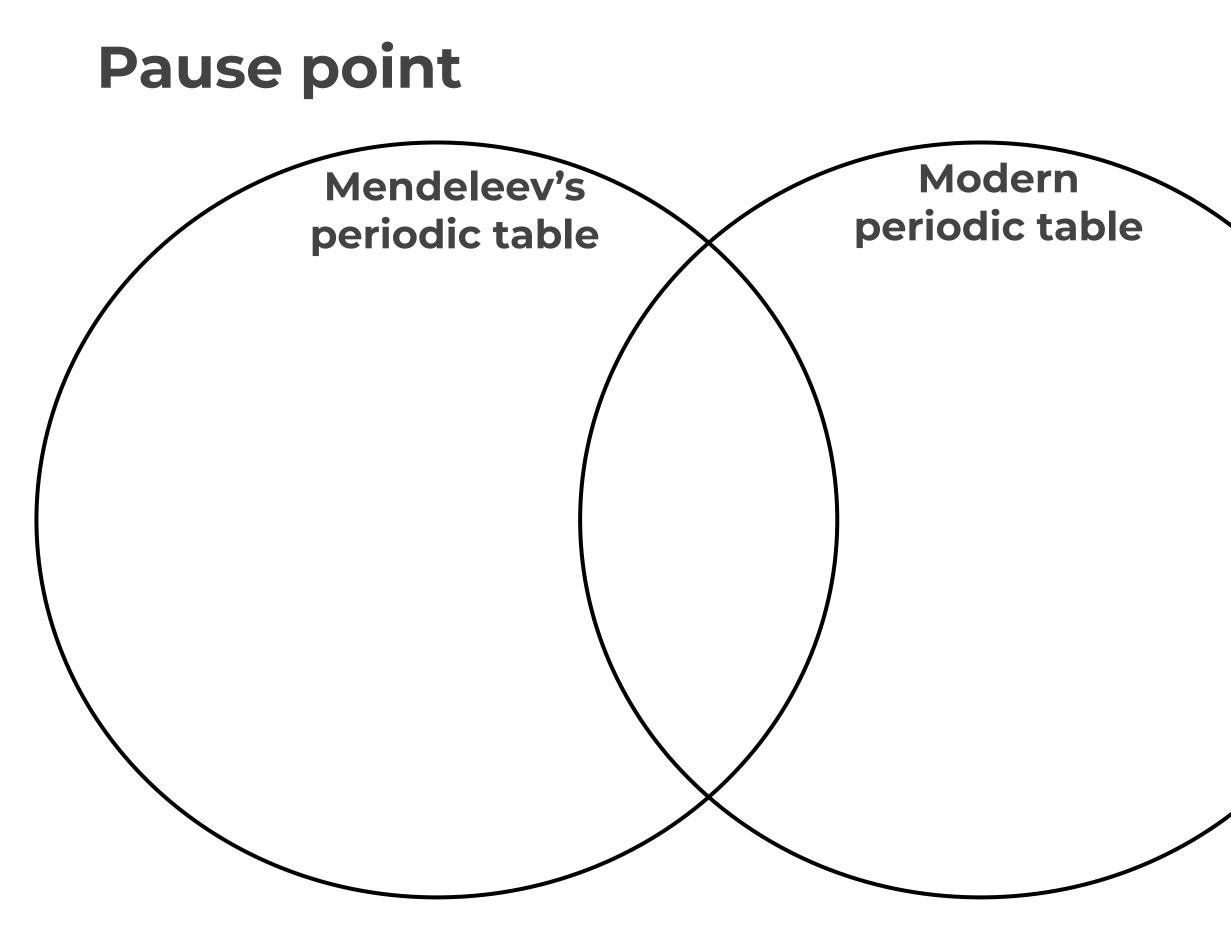


Independent practice

- 1. Where are metals found on the periodic table?
- 2. Where are all the non-metals found on the periodic table?
- 3. Complete the table:

Element name	Symbol	Metal or non-metal	Group number	Period number	
Sodium	Na	Metal	7	3	
Bromine		Non-metal			
	Ν		5	2	
	Ca				
			4	2	





Ordered elements using atomic <u>number</u>

Ordered elements using atomic <u>weight</u>

Contain groups

Groups have similar properties

Contains gaps

Contains every element



Independent practice

Compare the similarities and differences between the modern periodic table the Mendeleev's periodic table.

