Physics - Key stage 4 - Energy

## Insulating material required practical part 1 (Physics

 only) - worksheetDr Fishwick

## Exam style question

A student investigated how the thickness of insulation affects the cooling of a beaker of hot water.

The student wraps cotton wool around the beaker and measure the temperature at the start and after 15 minutes.

Describe a method the student could have used to obtain the results shown.

| Thickness of <br> cotton wool $/ 2$ <br> mm | Start <br> temperature <br> $/{ }^{\circ} \mathrm{C}$ | End <br> temperature <br> $/{ }^{\circ} \mathrm{C}$ | Temperature <br> change $/{ }^{\circ} \mathrm{C}$ |
| :--- | :--- | :--- | :--- |
| 0 | 90 | 65 | 25 |
| 2 | 90 | 67 | 23 |
| 4 | 90 | 69 | 21 |
| 6 | 88 | 69 | 19 |
| 8 | 88 | 71 | 17 |
| 10 | 90 | 75 | 15 |



Answers

## Points to include

- Wrap cotton wool around the beaker
- How the water has been heated (kettle / bunsen burner)
- Measuring the volume of water with a measuring cylinder.
- Pouring the water into the beaker
- Putting a lid on top of the beaker
- Measure initial and final temperature using thermometer
- Use stopwatch / stopclock to measure 10 minutes.
- Calculate temperature decrease
- Repeat with different thicknesses (specify 0-10 mm) of cotton wool
- Repeat with same volume of water - control variable

Level 3: The design/plan would lead to the production of a valid outcome. All key steps are identified and logically sequenced. 5-6

Level 2: The design/plan would not necessarily lead to a valid outcome. Most steps are identified, but the plan is not fully logically sequenced. 3-4

Level 1: The design/plan would not lead to a valid outcome. Some relevant steps are identified, but links are not made clear. 1-2

No relevant content: $\mathbf{O}$

## In lesson questions

Step

1. Put the small beaker inside the larger beaker.
2. Use the kettle to boil water. Put $\mathbf{8 0} \mathbf{c m}^{\mathbf{3}}$ of this hot water into the small beaker.

## Reason

To produce reliable results with the same size beaker This is the control variable and sets the starting point
3. Use a piece of cardboard as a lid for the large beaker. The cardboard must have a hole for the thermometer.
4. Put the thermometer through the hole in the cardboard lid so that its bulb is in the hot water. Leave it for 1 minute
5. Record the temperature of the water using a thermometer and start the stopwatch.
6. Record the temperature of the water every 3 minutes for 15 minutes.

## Imperative + Quantity + Equipment

## Design a method

Write a method to investigate the effect of thickness on cooling rate of water.
7. Wrap layers of insulating material around the beaker, holding it in place with a rubber band. Do not add insulating material to the bottom of the beaker.
2. Put ....... of ............ into the beaker.
3. Add a $\qquad$
4. Insert the
5. Record
6. .....
7. .....

- Independent variable - range, interval
- Dependent variable
- Reducing uncertainty - which is the biggest losses

Answers

## Step

1. Put the small beaker inside the larger beaker.

## Reason

To produce reliable results with the same size beaker
2. Use the kettle to boil water. Put $\mathbf{8 0} \mathbf{~ c m}^{\mathbf{3}}$ of this hot water into the small beaker.
3. Use a piece of cardboard as a lid for the large beaker. The cardboard must have a hole for the thermometer.
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6. Record the temperature of the water every 3 minutes for 15 minutes.

This is the control variable and sets
 transfer out the top of the beaker

So we can measure the dependent variable

Measuring
dependent variable

## Imperative + Quantity + Equipment

## Design a method

Write a method to investigate the effect of thickness on cooling rate of water.

1. Wrap two layers of insulating material around the beaker, holding it in place with a rubber band. Do not add insulating material to the bottom of the beaker.
2. Put $\mathbf{8 0} \mathbf{c m}^{3}$ of hot water into the beaker.
3. Add a lid to the beaker.
4. Insert the thermometer through the hole in the lid so that its bulb is in the hot water.
5. Record the temperature of the water using a thermometer and start the stopwatch.
6. Record the temperature of the water every $\mathbf{3}$ minutes for $\mathbf{1 5}$ minutes.
7. Repeat with $\mathbf{2}$ more layers of newspaper added.
8. Repeat until values for 0-10 layers have been collected.
