Chromatography

Chemistry - Key Stage 3

Particles - Lesson 12

Miss Mason



Recap

What is the separation technique of distillation used for?
To separate I_____ that have different b_____ p____.

Name 4 pieces of equipment necessary for the process of distillation.
C______f____, c_____t____t____c_____

What is a solute?
A substance that is able to d_____.

4. Describe what happens to the arrangement and movement of particles during 'condensing'. Condensing is the change of state that occurs when a ______ turns into a ______. In a gas, the particles have a ______ arrangement, they very rarely t ______ and they can move at all different _______. As the g_____ is c_____ and turns into a l______, the particles' store of k_______ energy decreases and they move ______ together. The forces of a _______ that were once between the particles are able to r______. The particles still have a _______ arrangement but they are now all t______ and are able to _______.

Place the following steps in the correct order

The solvent will travel up the filter paper, taking the soluble particles from the ink with it.

Place a lid on the beaker.

Draw a straight line using a pencil a few cm up from the bottom of the filter paper.

Submerge the filter paper into the water so that the pencil line is just above the water level.

Add the substance (e.g. ink) you want to test as a small dot on top of the pencil line.

The diffusion distance of each of the individual solutes depends on their solubility.

Add some solvent (water) to a beaker.



Copy and complete...

The baseline must be drawn in pencil because...

A lid must be placed on top of the beaker because...

Keywords: solute, solvent, soluble, insoluble, diffuse.



Chromatography is carried out on 3 pens to see which colours they contain.

- **1.** Which pen number matches up to substance Z?
- 2. Which colour can be found in all 4 substances?



